

## Effect of dielectric barrier discharge (DBD) plasma on the activity and structural changes of horseradish peroxidase

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### Abstract

As an emerging nonthermal technology, cold plasma has been used to inactivate endogenous enzymes that are responsible for enzymatic browning reaction of fruits and vegetables. This study aimed to investigate the inactivation effect of dielectric barrier discharge (DBD) plasma on horseradish peroxidase (HRP), a typical plant peroxidase. The results showed that DBD plasma caused inactivation of HRP in a time- and discharge power-dependent manner. The HRP activity decreased by 32.5, 50.6, 65.5, and 75.4%, respectively, after 2, 4, 6, and 8 min of exposure to DBD plasma at 57.6 W. The efficacy of DBD plasma for HRP inactivation was enhanced on increasing the discharge power from 6.0 to 57.6 W. Intrinsic fluorescence spectra showed that DBD plasma induced obvious structural changes in HRP. DBD plasma also caused fragmentation and carbonylation of HRP as well as the oxidative degradation of heme, which might be due to the reactive species in plasma. After DBD plasma exposure at 43.0 W for 8 min, there was approximately an 8.4-fold increase in surface hydrophobicity of HRP. After exposure to DBD plasma, the aggregation of HRP was observed by using atomic force microscopy analysis. In conclusion, DBD plasma causes structural changes and chemical modification of HRP, which may be responsible for the loss of enzymatic activity. These data contribute to the application of cold plasma in the control of enzymatic browning of food products during processing and storage.

**Keywords:** dielectric barrier discharge plasma; horseradish peroxidase; inactivation; mechanisms

### Introduction

Endogenous enzymes are widely found in fruit and vegetable products, such as polyphenol oxidase (PPO), peroxidase (POD), pectin methylesterase (PME), polygalacturonase (PG), and ascorbate oxidase (AO) (Hu *et al.*, 2013). Some endogenous enzymes may cause deterioration in the nutrients, flavor, texture, and color of foods during processing or preservation. POD and PPO are the main enzymes involved in oxidative deterioration reactions, resulting in the browning phenomenon and the loss of antioxidant activity of fruits and vegetables (Hu *et al.*, 2013). Lipoxygenase (LOX), a family of

nonheme-iron-containing fatty acid dioxygenases widely found in many plants (e.g., wheat, soybean, mung bean, and pea), can catalyze the oxidation of polyunsaturated fatty acids to yield hydroperoxides, leading to the loss of color, flavor, odor, and other processing qualities of plant-based products (Baysal and Demirdoven, 2007). Thus, these endogenous enzymes associated with quality decay should be inactivated to ensure the nutritional and sensory properties of foods. Traditionally, thermal treatments (e.g., hot water, steam, and microwave blanching) are extensively used to inactivate endogenous enzymes and decrease the microbial load in raw vegetables and fruits during processing operations (Deng *et al.*, 2019).

However, these thermal methods are known to cause detrimental effects on the nutritional, functional, and sensory properties (Petruzzi *et al.*, 2017). In addition, chemical preservatives (such as sulfur dioxide, sulfites, and citric acid) have also been used to inhibit enzyme activity, but chemical residues pose significant health risks to humans (Lien *et al.*, 2016). Therefore, emerging nonthermal technologies have been developed for the inactivation of deleterious endogenous enzymes and microorganisms in foods, such as high-pressure carbon dioxide, pulsed electric fields, high hydrostatic pressure, and ultrasound (Hu *et al.*, 2013; Wu *et al.*, 2020; Zhang *et al.*, 2019).

In the past few years, the application of cold plasma in the agriculture and food industry has been studied extensively (Liao *et al.*, 2019; Xiang *et al.*, 2018a). Plasma, the fourth state of matter, is an electrically neutral ionized gas, which generally consists of charged ions, electrons, metastable particles (such as excited atoms, molecules, and radicals), and photons (Niveditha *et al.*, 2021). Currently, cold plasma is mainly generated by electrical discharges, for example, dielectric barrier discharge (DBD), corona discharge, gliding arc discharge, spark discharge, and microwave discharge (López *et al.*, 2019). In the past few years, cold plasma has shown good application potential in the food industry, including food decontamination and the elimination of toxic chemical residues as well as food ingredients and food packaging modifications (Gavahian and Khaneghah, 2020; Gavahian *et al.*, 2020; Wang *et al.*, 2020; Xiang *et al.*, 2018b). For instance, Gavahian *et al.* (2019) found that DBD plasma treatment could effectively retain the textural and visual quality of shiitake mushrooms stored for 1 week.

In addition, cold plasma is also used to inactivate a range of endogenous food enzymes and modify the structure and properties of proteins. Following exposure to plasma-processed air for 10 min, the PPO and POD activities of fresh-cut apple were reduced by approximately 62 and 65%, respectively (Bußler *et al.*, 2017). Similar findings were also reported by Tolouie *et al.* (2018), who found that the residual activities of lipase and LOX of wheat germ were reduced to 25.03% and 49.98% after 25 min of treatment with DBD plasma at 24 kV. Peroxidases (PODs), a diverse group of oxidoreductases, are widely found in animals, plants, and microorganisms. PODs can catalyze the oxidation-reduction between  $H_2O_2$  and various reductants, and play an important role in many biological processes such as organ specificity, stress response, and pathogen infection (Hiraga *et al.*, 2001). According to the literature, class III plant peroxidase (EC 1.11.1.7) can catalyze the oxidation of endogenous phenolic compounds, consequently causing undesirable changes in the color, flavor, texture, and nutritional quality of plant-based foods during processing and storage (Burnette,

1977; Jang and Moon, 2011). Although the cold plasma-induced inactivation of POD has been well investigated (Bußler *et al.*, 2017; Pankaj *et al.*, 2013), the underlying mechanisms are not well clarified.

Based upon previous studies on cold plasma-induced inactivation of endogenous food enzymes, this study aimed to assess the effect of DBD plasma on the inactivation of horseradish peroxidase (HRP) isolated from horseradish (*Armoracia rusticana*) roots, a typical plant peroxidase. The effect of DBD plasma on the structure and properties of HRP was also investigated by using fluorescence, sodium dodecyl sulfate–polyacrylamide gel electrophoresis (SDS–PAGE), western blot assay, and atomic force microscopy (AFM). The results of the current study could provide a basis for applying cold plasma technology in enzymatic browning inhibition of fruits and vegetables.

## Materials and Methods

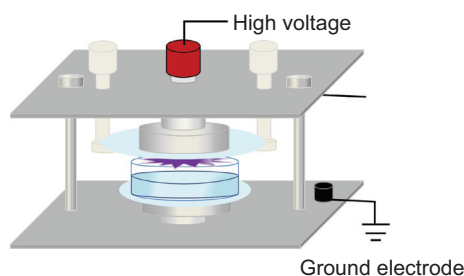
### Chemicals and reagents

Horseradish peroxidase (HRP, EC 1.14.18.1, RZ > 3.0) and guaiacol were provided by Shanghai Macklin Biochemical Co., Ltd. (Shanghai, China). Coomassie brilliant blue (CBB) R-250 and 1-anilinonaphthalene-8-sulfonate (ANS) were purchased from Aladdin Reagent Co., Ltd. (Shanghai, China). A protein carbonyl assay kit was purchased from Abcam (ab178020, Cambridge, UK). Polyvinylidene fluoride (PVDF) membranes with a pore size of 0.45  $\mu$ m and enhanced chemiluminescence (ECL) substrate were obtained from Millipore Corp. (Bedford, MA, USA).

### DBD plasma setup and treatment

The plasma system consists of a high voltage alternating current (AC) power source (CTP-2000 K), a DBD-50 plasma reactor, a voltage regulator, and a quartz dish (Nanjing Suman Electronics Co., Ltd., China). The two stainless steel electrodes (outer diameter = 53 mm) were covered with two 1-mm dielectric quartz plates (the diameters of the upper and lower dielectric discs were 100 and 90 mm, respectively). The distance between the quartz plates was approximately 15 mm. A quartz reactor (outer diameter = 70 mm, inner diameter 50 mm, depth = 4 mm, and height = 8 mm) was placed at the center of two dielectric discs (Figure 1). Air was used as the discharge gas (Li *et al.*, 2017). The discharge voltage and current were measured by an oscilloscope (DS1102E, Rigol).

HRP was dissolved in phosphate buffer (PB, 100 mmol/L, pH 7.0) at a concentration of 0.5 mg/mL. HRP solution



**Figure 1. Schematic diagram of the DBD plasma actuator.**

(2 mL) was placed in the quartz reactor and subjected to plasma treatment at different powers (6.0, 26.4, 43.0, and 57.6 W) for the indicated time period.

### Determination of enzymatic activity

The enzymatic activity of HRP was determined by the method of Wang *et al.* (2017). After each treatment, 90  $\mu$ L of guaiacol solution (20 mmol/L in ultrapure water) was mixed with 200  $\mu$ L of  $H_2O_2$  solution (0.25%, v/v) and 10  $\mu$ L of HRP solution (0.5 mg/mL). Then, the absorbance was measured at 470 nm for 180 s using a Tecan Spark 20M multimode microplate reader (Tecan Group Ltd., Männedorf, Zürich, Switzerland). One unit of enzymatic activity (U) was defined as an increase of 0.001 units of absorbance per minute under the assay conditions. The relative activities of HRP were obtained with the following formula:

$$\text{Residual activity} = \frac{A}{A_0} \times 100\% \quad (1)$$

where A is the specific activity of HRP treated with DBD plasma and  $A_0$  is the initial activity of HRP solution without DBD plasma treatment.

### Analysis of intrinsic fluorescence spectra

Intrinsic fluorescence spectral analysis was conducted using a fluorescence spectrophotometer (F-7100, Hitachi High-Technologies Corp., Tokyo, Japan). In brief, HRP solutions (0.5 mg/mL in 100 mmol/L PB, pH 7.0) were excited at 280 nm, and the emission wavelength ranged from 290 to 340 nm at a scan speed of 240 nm/min. The emission and excitation slit widths were both adjusted to 5 nm. Fluorescence intensity is measured in arbitrary units (A.U.).

### Analysis of protein oxidative damage

HRP solutions (0.5 mg/mL) were treated with DBD plasma at 43 W for 0, 2, 4, 6, and 8 min, respectively.

Thereafter, the oxidative damage of HRP was assessed by SDS–PAGE (Dong *et al.*, 2021). In brief, each sample was mixed with 4  $\times$  sample loading buffer and heated in boiling water for 5 min. Protein samples were separated on 10% SDS–polyacrylamide gel and the gel was stained with CBB R-250 solution for 1 h with gentle agitation. Following the destaining step, the gel was visualized by using a ChemiDoc XRS<sup>+</sup> system (Bio–Rad Laboratories, Hercules, CA, USA). The levels of carbon-ylated HRP were detected using a protein carbonyl assay kit (ab178020, Abcam) according to the manufacturer's protocols. The immunoreactive proteins were visualized using a chemiluminescence protocol, and the images were recorded with a ChemiDoc XRS<sup>+</sup> system (Bio–Rad).

### Determination of surface hydrophobicity

The ANS probe was used to determine the surface hydrophobicity index (SHI) of HRP before or after exposure to DBD plasma (Pustulka *et al.*, 2020). The HRP solution (4 mL, 0.2–1.0 mg/mL) was added to 20  $\mu$ L of ANS stock solution (8 mmol/L in PB of 100 mmol/L, pH 7.0), and then mixed well. After 15 min of incubation in the dark, the relative fluorescence intensity was measured using a HITACHI F-7100 fluorescence spectrometer (Tokyo, Japan) at excitation and emission wavelengths of 390 nm and 470 nm, respectively. The excitation and emission slit widths were both 5 nm. The slope of the relative fluorescence intensity versus protein concentration plot was calculated by linear regression, and the slope of the initial segment was reported as the SHI ( $S_0$ ).

### Atomic force microscopy analysis

The changes in the molecular microstructure of HRP were further assessed by AFM. After treatment with DBD plasma at 43.0 W for 0, 4, and 8 min, 10  $\mu$ L of HRP solution (0.5 mg/mL) was dropped onto a freshly cleaved mica surface and air-dried in a clean hood for 1 h at ambient temperature. Thereafter, the samples were measured by AFM (NanoMan VS, Bruker, Karlsruhe, Germany) in the tapping mode (Wang *et al.*, 2017). Samples were scanned with a force constant of 2 N/m, a scan rate of 2 Hz, and a step size of 3  $\times$  3  $\mu$ m. The obtained topographical images were analyzed with NanoScope Analysis 1.7 software (Bruker).

### Analysis of heme damage

HRP solutions (0.5 mg/mL) were treated with DBD plasma at 43.0 W for 0, 2, 4, 6, and 8 min. The absorption spectra of HRP solutions were scanned in the range of 250 to 550 nm with a UV-Vis spectrophotometer

(1800PC, Macylab Instruments Inc., Shanghai, China). The fluorescent heme degradation products were analyzed according to a previously described method (Nagababu and Rifkind, 2004). HRP solutions (0.5 mg/mL in 100 mmol/L PB, pH 7.0) were excited at 330 nm, and the emission was collected from 340 to 600 nm at a scan speed of 240 nm/min. Both the excitation and emission slit widths were set to 5 nm.

### Statistical analysis

All analyses were carried out in triplicate and experimental results are expressed as the mean  $\pm$  standard deviation (SD). The data were analyzed using one-way analysis of variance (ANOVA), and the differences between the means were tested by using the Fisher's least significant difference (LSD) test at  $p < 0.05$ . Statistical analyses were conducted using SPSS Statistics for Windows, Version 22.0 (IBM Corp., Armonk, NY, USA).

## Results and Discussion

### Inactivation of HRP induced by DBD plasma

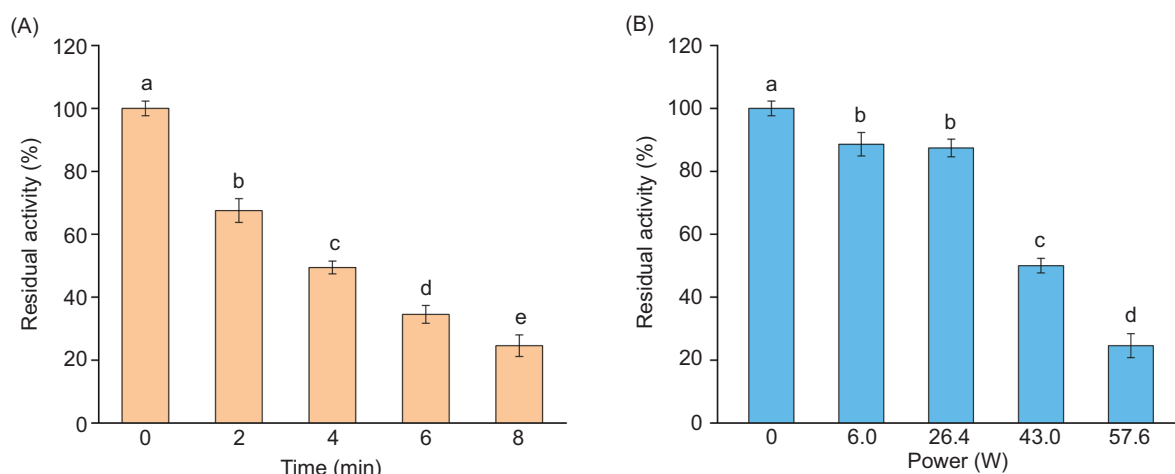
HRP isolated from horseradish tissue was used in this work. Horseradish (*Armoracia rusticana*) is a perennial herb of the *Brassica* mustard family and is cultivated mainly for the culinary value of its pungent roots (Veitch, 2004). *A. rusticana* roots provide an abundant source of peroxidase, and thus, they are the main source of industrial HRP. The inactivation of HRP by DBD plasma is presented in Figure 2. The initial activity of HRP was 228 U/mL, which decreased continuously as a function of the DBD plasma exposure time. After 2, 4, 6, and 8 min

of exposure to DBD plasma at 57.6 W, the HRP activity decreased by 32.5, 50.6, 65.5, and 75.4%, respectively (Figure 2A). Similar findings were reported in the studies of Wang *et al.* (2021), in which the residual activity of POD extracted from litchi pericarp decreased to 47.16% after DBD plasma treatment for 10 min. Chutia *et al.* (2019) reported that the DBD plasma treatment time was a significant factor for the inactivation of PPO and POD in tender coconut water. Illera *et al.* (2019) also reported that spark discharge plasma could effectively inactivate PPO in a pure enzymatic solution or cloudy apple juice in a time-dependent manner.

The discharge power also played a critical role in DBD-plasma-mediated HRP inactivation. The residual activities of HRP steadily decreased with increasing discharge power of DBD reaction. The HRP activity decreased by 11.4, 12.7, 50, and 75.4%, respectively, after 8 min of exposure to DBD plasma at 6.0 W, 26.4 W, 43.0 W, and 57.6 W, respectively (Figure 2B). Similar findings were also previously reported by Chutia *et al.* (2019), who found that greater inactivation efficacy of DBD plasma on PPO and POD in tender coconut water was observed at a higher voltage. This may be because more reactive species were generated during gas-phase plasma discharge supplied with a higher power, which is responsible for the plasma-elicited inactivation effects (Pai *et al.*, 2018).

### Intrinsic fluorescence spectra of DBD plasma-treated HRP

Proteins and peptides exhibit intrinsic fluorescence when excited with UV light, which originates from the three amino acid residues with aromatic side chains:



**Figure 2.** Residual activity of HRP after exposure to DBD plasma. (A) HRP solutions (0.5 mg/mL) were treated with DBD plasma at 57.6 W for the indicated time periods. (B) HRP solutions (0.5 mg/mL) were treated with DBD plasma at the indicated power for 8 min. Values with different letters are significantly different based on the LSD test at  $p < 0.05$ .



tryptophan (Trp), tyrosine (Tyr), and phenylalanine (Phe). Among the three aromatic amino acids, Trp is regarded as the dominant intrinsic fluorophore of proteins due to its highest quantum yield and largest extinction coefficient (Ghisaidoobe and Chung, 2014). Hence, the intrinsic fluorescence of proteins has been widely used to probe changes in protein structure and therefore function. As depicted in Figure 3, the fluorescence intensity of HRP gradually decreased after exposure to DBD plasma. The initial intrinsic fluorescence of the HRP solution was 6696 A.U. (Figure 3A). After DBD plasma treatment at 43.0 W for 8 min, the intrinsic fluorescence of the HRP solution decreased to 2553 A.U. (Figure 3A). DBD-plasma-induced decrease in the intrinsic fluorescence of HRP was also affected by the discharge power (Figure 3B). The obvious changes in the fluorescence intensity of HRP might be due to the structural changes in HRP induced by DBD plasma.

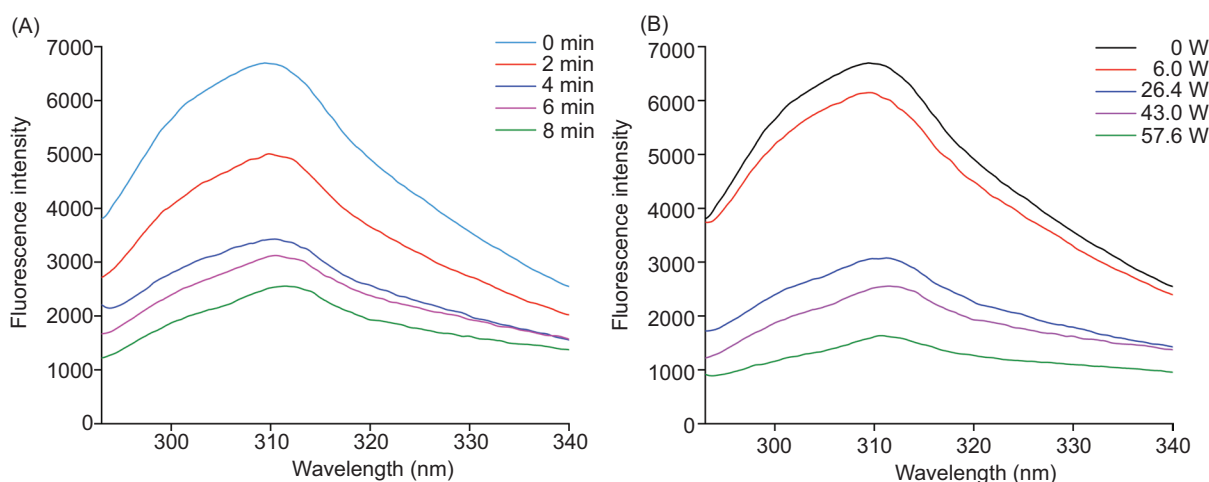
The largest emission peak ( $\lambda_{\max}$ ) of the native HRP was at 309.4 nm, suggesting that Trp is buried in a “nonpolar” environment (Vivian and Callis, 2001). Slight changes were observed in  $\lambda_{\max}$  after DBD plasma treatment. After DBD plasma treatment at 43.0 W for 8 min, the  $\lambda_{\max}$  of HRP was red-shifted to 311.4 nm (Figure 3A). These data indicated that Trp residues of HRP moved to the outside of the protein molecules, causing an increase in the polarity of their local microenvironment (Jiang *et al.*, 2014). These data reflect the changes in the three-dimensional structures and conformation of HRP after exposure to DBD plasma, which may be responsible for the loss of enzyme activity.

### DBD plasma caused oxidation and carbonylation of HRP

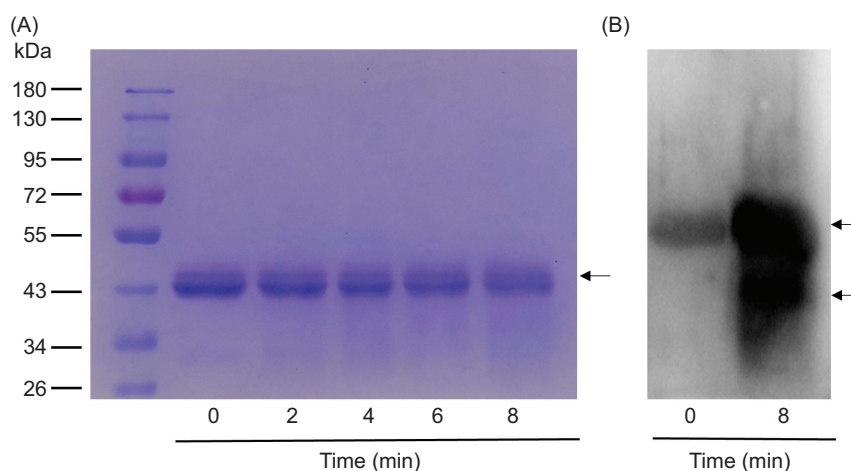
The oxidative degradation of HRP induced by DBD plasma was monitored by SDS–PAGE under reducing

conditions. As depicted in Figure 4A, a distinct HRP band was observed approximately at 43 kD. The intensity of HRP bands decreased in a time-dependent manner after DBD plasma treatment (Figure 4A). These data are consistent with previous findings (Ng *et al.*, 2020; Venkataratnam *et al.*, 2020). For instance, SDS–PAGE analysis indicated that the band intensity of casein declined dramatically with an increasing treatment time of spark discharge and glow discharge plasma (Ng *et al.*, 2020). Similar findings were also reported by Venkataratnam *et al.* (2020), who found that cold plasma caused degradation of Ara h1 and Ara h2 with SDS–PAGE, two major peanut allergens. The decrease in band intensity following DBD plasma treatment may be due to the oxidative damage of HRP. During gas discharge, various active molecules are generated, such as charged particles, energetic ions, reactive species (e.g., ozone, hydroxyl radicals, superoxide, peroxide, and nitric oxide radicals), and ultraviolet (UV) radiation (Gaunt *et al.*, 2006). These species may lead to the cleavage of peptide bonds, resulting in the fragmentation of HRP as indicated by the arrow in Figure 4A.

The formation of carbonyl groups is the most widely used marker of protein oxidation. In this work, carbonylated HRP was measured by western blot assay using a protein carbonyl assay kit (ab178020, Abcam). As shown in Figure 4B, obvious carbonylated HRP bands were observed after exposure to DBD plasma at 43.0 W for 8 min. Similar findings were also observed in the studies of Attri *et al.* (2015) and Panpipat and Chaijan (2020), who stated that the carbonyl contents of hemoglobin, myoglobin, and actomyosin from threadfin bream were remarkably increased after being subjected to cold plasma. Reactive carbonyl derivatives (e.g., aldehydes, ketones, and lactams) can be generated through oxidative



**Figure 3.** Fluorescence emission spectra of HRP after DBD plasma treatment. (A) HRP solutions were treated with DBD plasma at 43.0 W for the indicated time. (B) HRP solutions were treated with DBD plasma at 6.0, 26.4, 43.0, and 57.6 W for 8 min, respectively.



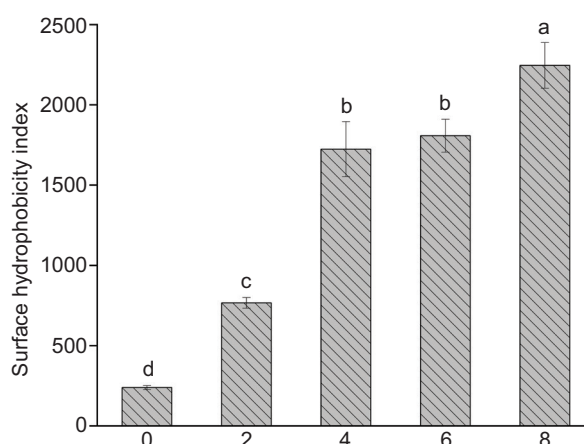
**Figure 4.** Oxidative damage of HRP after DBD plasma treatment at 43.0 W for the indicated time. (A) SDS–PAGE patterns of HRP treated with DBD plasma. (B) Analysis of carbonylated HRP with western blot assay.

cleavage of the protein backbone or modification of amino acid side chains induced by free radicals (Fedorova *et al.*, 2014). As indicated by the arrow in Figure 4B, the immunoreactive bands were also observed in the lower molecular weight region of the western blot, suggesting that the oxidative degradation of HRP was caused by the active molecules in DBD plasma. These data reveal that the reactive species in DBD plasma may induce modification of amino acid side chains or backbone cleavage in HRP molecules, which may subsequently result in the loss of enzymatic activity.

#### Effect of DBD plasma on the surface hydrophobicity of HRP

Surface hydrophobicity is an important surface-related property of proteins. The effect of DBD plasma on the surface hydrophobicity of HRP is shown in Figure 5. The  $S_0$  values for HRP were significantly increased as a function of DBD plasma treatment time ( $p < 0.05$ ). After exposure to DBD plasma at 43.0 W for 8 min, the  $S_0$  value of HRP increased to 2245.75, which was significantly higher than that of the native protein molecule. These data were in agreement with those reported in previous works. Segat *et al.* (2015) found that the surface hydrophobicity of whey protein isolate increased after exposure to DBD plasma for 1 to 30 min but decreased remarkably as the treatment time increased to 60 min.

The surface hydrophobicity of proteins is determined by various factors, such as the size, shape, amino acid composition and sequence, and the intramolecular or intermolecular cross-linking of proteins (Jiang *et al.*, 2015). The increased surface hydrophobicity might be due to the structural unfolding of HRP after exposure to DBD plasma. During DBD plasma treatment, the hydrophobic

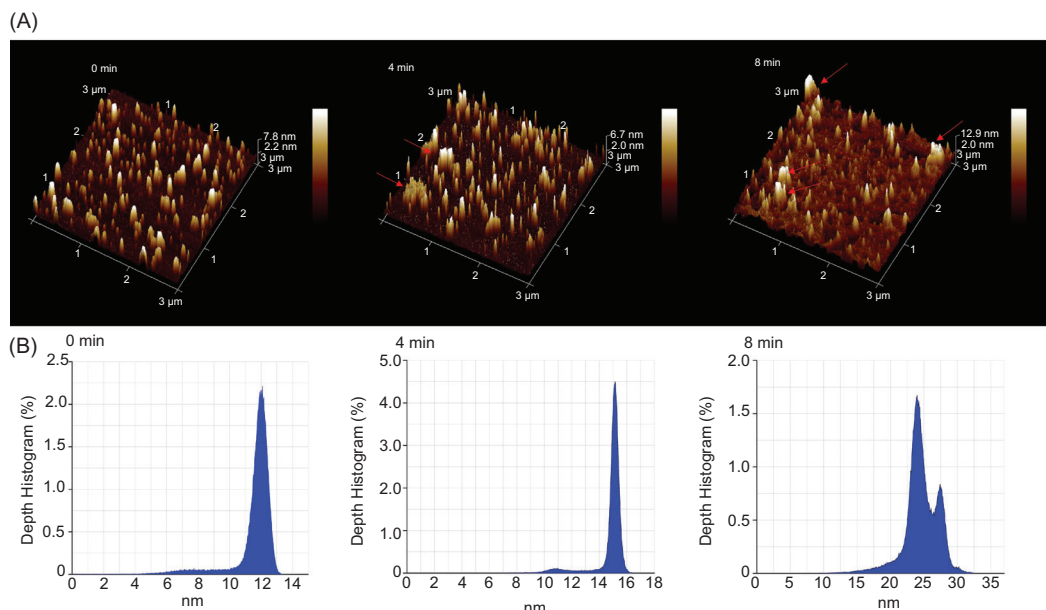


**Figure 5.** Surface hydrophobicity of HRP after exposure to DBD plasma at 43.0 W for the indicated time periods. Means followed by different letters are significantly different ( $p < 0.05$ ) according to the LSD test.

amino acids buried within the protein molecule might become more hydrophobic on the surface of the protein, where they were more accessible to the ANS probes (Segat *et al.*, 2015). In addition, DBD-plasma-induced oxidative degradation of HRP may also contribute to changes in the surface hydrophobicity (Jiang *et al.*, 2015).

#### Particle size analysis of HRP by AFM

The AFM images of HRP before and after DBD plasma treatment are displayed in Figure 6A. The native HRP particles on the mica surface were uniform with an average diameter of 106 nm, and the average height was 12 nm. As depicted in Figure 6A, particulate aggregates of HRP proteins were observed after exposure to DBD plasma. After DBD plasma treatment at 43.0 W for



**Figure 6.** AFM topography images (A) and height distribution of HRP (B) before or after DBD plasma treatment at 43.0 W for 0, 4, and 8 min.

4 and 8 min, the average diameter of HRP particles was increased to 147 and 229 nm, respectively (Figure 6A). In addition, as shown in Figure 6B, after exposure to DBD plasma, the average height was remarkably increased, and the average height was increased to 15 and 25 nm after DBD plasma treatment at 43.0 W for 4 and 8 min, respectively.

Similar aggregation was also reported for HRP, thaumatococcus-like proteins, and PPO treated with pulsed light (Wang *et al.*, 2017), high-pressure carbon dioxide (Li *et al.*, 2014), and heat (Zhou *et al.*, 2017). AFM analysis revealed that the aggregation of enzyme proteins increased after exposure to pulsed light (Wang *et al.*, 2017). DBD plasma caused denaturation and aggregation of HRP, which might contribute to the loss of enzymatic activity (Zhou *et al.*, 2017).

### DBD plasma caused damages to heme

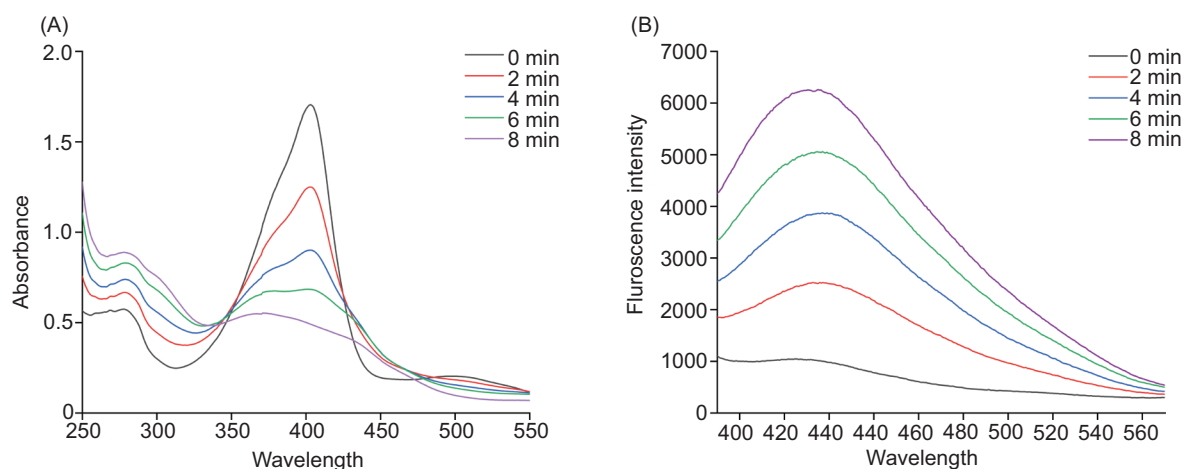
HRP has two metal centers, one heme group and two calcium atoms. The heme group is located at the catalytic reaction center of HRP and plays an important role in maintaining the three-dimensional structure of the enzyme (Veitch, 2004). The characteristic structures of HRP before or after plasma exposure were determined by measuring UV-visible spectroscopy. As shown in Figure 7A, native HRP showed a Soret band at 403 nm, corresponding to the presence of the heme group. After exposure to DBD plasma, the Soret band at 403 nm significantly increased in a time-dependent manner. This

indicates that DBD plasma causes extensive conformational changes in the protein molecule.

The detrimental effect of DBD plasma on the heme group was also analyzed by measuring the fluorescent heme degradation products. As shown in Figure 7B, native HRP displayed the largest emission peaks at 425 nm with a fluorescence intensity of 1045 A.U. However, DBD plasma caused remarkable increases in the fluorescence intensity of HRP. After exposure to DBD plasma at 43.0 W for 8 min, the fluorescence intensity of HRP increased to 6259 A.U., while no shift was observed. The elevated fluorescence intensity may be due to the oxidative degradation of heme caused by RONS, such as  $H_2O_2$ , in the air DBD plasma. Previous studies show that fluorescent products are formed during the reaction of heme and hemin with hydroperoxides, including  $H_2O_2$ , cumene hydroperoxide, and tertiary butyl hydroperoxides. In addition, the release of iron from the heme moiety is also observed from the reaction of heme and hydroperoxides (Nagababu and Rifkind, 2004). These data suggest that the chemically active species causes extensive conformational changes in the HRP molecule, consequently resulting in loss of enzymatic activity.

### Conclusions

The present study reported the inactivation effect and mechanisms of DBD plasma on HRP. A significant loss of HRP activity was observed after DBD plasma treatment, which was affected by the exposure time



**Figure 7.** Detrimental effects of DBD plasma treatment on the heme group in HRP. (A) UV-visible absorption spectra of HRP solution (0.5 mg/mL) after DBD plasma treatment at 43.0 W for the indicated time periods. (B) Formation of fluorescent heme degradation products during the reaction of HRP with DBD plasma at 43.0 W for 0, 2, 4, 6, and 8 min, respectively.

and discharge power. DBD plasma caused remarkable changes in the three-dimensional structure or conformation of the protein. In addition, fragmentation and carbonylation of HRP as well as the oxidative degradation of heme were also observed after exposure to DBD plasma, which might be due to the reactive species in plasma. AFM analysis of HRP suggested that DBD plasma induced aggregation of HRP. These data indicate that the HRP inactivation induced by DBD plasma is mainly due to the structural changes and oxidative modification of the protein molecule. In summary, cold plasma is a promising new technology to inhibit the enzymatic browning of fruits and vegetables. However, cold plasma technology is still in its infancy, and several technological challenges need to be solved before its industrial application, such as the scale-up of plasma-generating devices and the optimization of processing parameters. In addition, the influences of cold plasma on the quality of food products and human health should be well evaluated before this technology can be fully utilized.

## Conflicts of Interest

The authors declare no conflicts of interest.

## Acknowledgments

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